











The International Scientific Conference "Natural and Synthetic Polymers for Medical and Technical Purposes"

Book of abstracts

Minsk, Belarus, April 27 – 29, 2022



The conference is dedicated to the memory of the outstanding Belarusian scientist and teacher, academician of the NAS of Belarus

Kaputsky Fyodor Nikolaevich (1930 - 2017)





BELARUSIAN STATE UNIVERSITY БЕЛОРУССКИЙ ГОСУДАРСТВЕННЫЙ УНИВЕРСИТЕТ RESEARCH INSTITUTE FOR PHYSICAL CHEMICAL PROBLEMS

PROBLEMS

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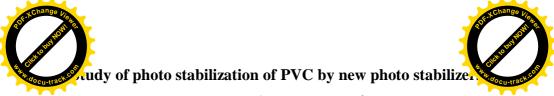
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Thermal destruction of polyvinyl chloride occurs by sequential removal of hydrogen chloride (HCl), which is called dehydrochlorination, with the formation of long polyenes, which as a result cause color change, deterioration of mechanical properties and a decrease in chemical resistance.

We have synthesized photostabilizers based on derivatives of crotonic aldehyde and the corresponding d-metals, which are used as PVC photostabilizer. The stabilization of stabilized PVC samples was studied with UV radiation over time.

The change in the properties of samples from time to time was studied by the IR spectroscopic method. In the IR spectrum of the presented samples, a noticeable change in the region of $3435~\rm cm^{-1}$ occurs during aging; in this connection, the band grows to $1730~\rm cm^{-1}$ and the intensity of the bands decreases in the region of 1632, 1125, 1077, $967~\rm cm^{-1}$, and a number of others. A wide band of stretching vibrations in the region of $3000{-}4000~\rm cm^{-1}$ proves the appearance of hydroxyl-OH groups, and also corresponds to various bonded types of hydrogen bond, and a peak in the region of $1735~\rm cm^{-1}$ corresponds to deformation vibrations of carbonyl C = O groups, which forms in the process various functional groups, which are usually also associated with a hydrogen bond.

Thus, the studied data confirmed that within 3000 hours during the aging process, structural changes are observed (a decrease in the concentration of C–O–C groups, an increase in the content of CO and OH groups, as well as the formation and opening of C = C bonds with the formation of cross-links) leads to a slight increase in strength characteristics, proves the crosslinking process during radiation (tensile strength and elastic modulus) and the stiffness of the material, a decrease in its elasticity (relative elongation at break), and deterioration of decorative character eristik (reduced - contact angle, transmittance increases discoloration and gloss).

The kinetics of the mass loss of PVC from the heating temperature in the temperature range $290-770\,^{\circ}\mathrm{C}$ under consideration is associated with various processes: the removal of hydrogen chloride, decomposition with the participation of oxygen, the release of volatile substances, etc. The mass loss in the TGA curve with increasing temperature is due to the continuation of the destruction of PVC. This part of the process is accompanied by an exothermic effect.

Thus, on the basis of the obtained experimental data, it can be concluded from the kinetics of thermooxidative degradation of PVC in the temperature range $290-770\,^{\circ}\text{C}$ that, firstly, the removal of hydrogen chloride in the initial sample occurs in the range $300-600\,^{\circ}\text{C}$, and for stabilized samples - in the range of $500-800\,^{\circ}\text{C}$.